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References

- ERIKSSON, B. (1982). Acta Chem. Scand. Ser. A, 36, 186-188. ERIKSSON, B., LARSSON, L. O., NIINISTÖ, L. & VALKONEN, J. (1980). Inorg. Chem. 19, 1207-1210.
- 5, 349-352.

International Tables for X-ray Crystallography (1962), Vol. III. Birmingham: Kynoch Press. (Present distributor D. Reidel, Dordrecht.)

MILINSKI, N., RIBÁR, B. & SATARIĆ, M. (1980). Cryst. Struct. Commun. 9, 473-477.

RIBÁR, B., MILINSKI, N., BUDOVALČEV, Ž. & KRSTANOVIĆ, I. (1980). Cryst. Struct. Commun. 9, 203-206.

ROGERS, D. J., TAYLOR, N. J. & TOOGOOD, G. E. (1983). Acta Cryst. C39, 939-941.

Acta Cryst. (1986). C42, 1452-1454

Structures of (+)-8,10- and (+)-9,10-Dibromocamphor*

BY STEVEN J. RETTIG AND JAMES TROTTER

Department of Chemistry, University of British Columbia, Vancouver, BC, Canada V6T 1Y6

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Abstract. $C_{10}H_{14}Br_2O$, $M_r = 310.03$, Mo Ka, $\lambda a_1 =$ 0.70930 Å, T = 295 K; 8,10 isomer (1): monoclinic, $P2_1$, a = 7.438(1), b = 7.293(1), c = 10.177(1) Å, scan width $(0.65 + 0.35 \tan \theta)^\circ$ at $1.3-10^\circ \min^{-1}$, $\beta = 95.689 (5)^{\circ}, V = 549.3 (1) \text{ Å}^3, Z = 2, D_x = 1.874 (1) \text{ g cm}^{-3}, \mu = 73 \text{ cm}^{-1}, F(000) = 304, R = 1.874 (1) \text{ g cm}^{-3}, \mu = 73 \text{ cm}^{-1}, F(000) = 304, R = 1.874 \text{ cm}^{-1}$ 0.051 for 891 reflections; 9,10 isomer (2): orthorhombic, $P2_12_12_1$, a = 10.607(1), b = 10.610(1), c= 9.8630 (3) Å, V = 1109.98 (12) Å³, Z = 4, $D_x = 1.855$ (1) g cm⁻³, $\mu = 72$ cm⁻¹, F(000) = 608, R = 1.855 (1) g cm⁻³, $\mu = 72$ cm⁻¹, F(000) = 608, R = 1.855 (1) g cm⁻³, $\mu = 72$ cm⁻¹, F(000) = 608, R = 1.855 (1) g cm⁻³, $\mu = 72$ cm⁻¹, F(000) = 608, R = 1.855 (1) g cm⁻³, $\mu = 72$ cm⁻¹, F(000) = 608, R = 1.855 (1) g cm⁻³, $\mu = 72$ cm⁻¹, F(000) = 608, R = 1.855 (1) g cm⁻³, $\mu = 72$ cm⁻¹, F(000) = 608, R = 1.855 (1) g cm⁻³, $\mu = 72$ cm⁻¹, F(000) = 608, R = 1.855 (1) g cm⁻³, $\mu = 72$ cm⁻¹, F(000) = 608 (12) F(00) = 6080.047 for 805 reflections. The geometries and dimensions of the camphor ring systems are similar to those in related molecules; the C(1)-C(7)-C(4) bridgehead angles are 94.0(7) and $92.8(9)^\circ$, and the mean C-Br bond distances are 1.968 (6) and 1.960 (9) Å.

Experimental. Colourless crystals, dimensions $0.25 \times$ 0.30×0.21 mm for 8,10-dibromo isomer (1) [0.48 × 0.48×0.28 mm for 9,10-dibromo isomer (2)], crystal faces $\{001\}$, $(\overline{1}10)$, (100), (101) [$\{100\}$, $\{011\}$]; Enraf-Nonius CAD-4F diffractometer; lattice parameters from 25 reflections with $\theta = 20-26^{\circ}$ [18-



* (1R,4S,7R)- and (1R,4S,7S)-1,7-bis(bromomethyl)-7-methylbicyclo[2.2.1]heptan-2-one.

* $U_{eq} = \frac{1}{3}$ trace of diagonalized U tensor.

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24°]. Intensities for $\theta \leq 30^{\circ}$ [25°]; *hkl*: 0 to 10, 0 to 10, -14 to 14 [0 to 13, 0 to 13, 0 to 12]; ω -2 θ scan, ω

Table 1. Final positional (fractional, $\times 10^4$, Br $\times 10^5$) equivalent isotropic thermal parameters and $(U \times 10^3 \text{ Å}^2)$ with e.s.d.'s in parentheses

	x	У	z	Ueu*
8,10-Dibi	omocamphor			
Br(1)	9521 (16)	40000	103038 (9)	60
Br(2)	21598 (17)	36412 (29)	36724 (9)	72
0	2248 (11)	331 (11)	6044 (8)	56
C(1)	2994 (12)	3586 (14)	6515 (8)	33
C(2)	2906 (13)	1523 (16)	6761 (10)	36
C(3)	3833 (15)	1206 (16)	8144 (10)	38
C(4)	4418 (15)	3119 (15)	8571 (10)	85
C(5)	5970 (12)	3707 (19)	7761 (12)	54
C(6)	5018 (12)	3927 (19)	6365 (9)	45
C(7)	2822 (11)	4311 (14)	7928 (8)	29
C(8)	984 (12)	3861 (21)	8383 (7)	43
C(9)	3086 (17)	6387 (18)	8094 (11)	47
C(10)	1629 (14)	4292 (20)	5467 (8)	46
9.10-Dibr	omocamphor			
Br(1)	40245 (17)	33271 (17)	97060 (15)	68
Br(2)	25380 (20)	29256 (14)	30022 (13)	58
0	590 (9)	1241 (8)	4951 (12)	53
C(1)	2490 (14)	2231 (9)	5858 (10)	34
C(2)	1349 (13)	1337 (12)	5843 (17)	30
C(3)	1390 (13)	623 (14)	7163 (17)	46
C(4)	2564 (16)	1161 (11)	7853 (14)	46
C(5)	3710 (13)	667 (16)	7102 (20)	52
Ció	3643 (12)	1320 (12)	5727 (16)	38
C(7)	2518 (16)	2574 (11)	7413 (12)	36
C(8)	1360 (13)	3289 (15)	7885 (17)	49
C(9)	3694 (13)	3301 (14)	7748 (13)	40
C(10)	2442 (14)	3362 (12)	4915 (11)	39

extended 25% on each side for background measurement; three standard reflections (average 3% decay); Lp and intensity absorption corrections, transmission factors 0.139-0.287 [0.055-0.170]; 1715 [1458] independent reflections measured, 891 [805] with $I \ge 3\sigma(I), \ \sigma^2(I) = S + 4(B_1 + B_2) + (0.04S)^2, \ S = \text{scan},$ B_1 , B_2 = background counts. Structure by Patterson and Fourier methods; refined by full-matrix least squares on F; H atoms from difference map, but fixed in calculated positions in final cycles; $w = 1/\sigma^2(F)$; scattering factors, including anomalous dispersion for Br, from International Tables for X-ray Crystallography (1974); locally written or locally modified versions of standard computer programs; final R = 0.051, wR = 0.058 for 891 reflections, S = 2.49,117 parameters [0.047, 0.055 for 805 reflections, 1.68, 119 parameters, including extinction correction, g $= 2.54 (4) \times 10^{-4}$; R = 0.109 for all 1715 reflections [0.099 for all 1458 reflexions]; $\Delta/\sigma = 0.01$ (mean),

Table 2. Bond lengths (Å) and angles (°) with e.s.d.'s in				
parentheses				

	8,10-Dibromocamphor $(n = 8)$		9,10-Dibromocamphor (n = 9)	
	Uncorr		Uncorr	
$\mathbf{P}_{\mathbf{r}}(1) = \mathbf{C}(\mathbf{r})$	1.060 (9)	1.065	1.062 (12)	1.069
Br(1) = C(n) Br(2) = C(10)	1.065 (0)	1.905	1.905 (15)	1.051
$D_{1}(2) = C(10)$	1.905 (9)	1.211	1.109 (11)	1 201
C(1) $C(2)$	1.207(13)	1.524	1.528 (18)	1.542
C(1) - C(2)	1.53(2) 1.549(12)	1.554	1.564 (18)	1.570
C(1) = C(0)	1.546(13)	1.555	1.576 (16)	1.500
C(1) = C(1)	1.490 (11)	1.407	1.510(10)	1.522
C(1) - C(10)	1.522 (13)	1.520	1.517 (13)	1.512
C(2) = C(3)	1.522(14)	1.519	1.520 (22)	1 524
C(3) = C(4)	1.51(2)	1.510	1.529 (20)	1.534
C(4) = C(3)	$1 \cdot 54(2)$	1.549	1.518 (21)	1.524
C(4) = C(7)	1.503 (14)	1 520	1.502 (15)	1.507
C(3) - C(0)	1.532 (14)	1.536	$1 \cdot 524 (23)$	1.529
C(7) = C(8)	1.522(13)	1.527	1.510 (18)	1.507
$C(\eta - C(\theta))$	1.55 (2)	1.239	1.303 (19)	1.307
C(2)-C(1)-C(6)	103-4	4 (8)	103.5 (8	;)
C(2)-C(1)-C(7)	100.	1 (8)	99.6 (1	1)
C(2)-C(1)-C(10)	114.	9 (9)	117.0 (1	1)
C(6)C(1)C(7)	102.0	0 (7)	102.0 (1	1)
C(6)-C(1)-C(10)	118-	3 (8)	117.7 (1	1)
C(7)C(1)C(10)	115.	7 (8)	114.5 (9)
O - C(2) - C(1)	129.	1 (10)	126-1 (1	3)
O - C(2) - C(3)	124.	7 (10)	127.7 (1	2)
C(1)-C(2)-C(3)	106-3	2 (8)	106.2 (1	2)
C(2)-C(3)-C(4)	102.	5 (9)	102.7 (1	1)
C(3)-C(4)-C(5)	108-1	2 (10)	107.8 (1	1)
C(3)-C(4)-C(7)	102.0	0 (8)	102.1 (1	2)
C(5)-C(4)-C(7)	101.	5 (8)	102.7 (1	2)
C(4) - C(5) - C(6)	102.	9 (8)	103.9 (1	1)
C(1)-C(6)-C(5)	104.	8 (7)	104-1 (1	1)
C(1)-C(7)-C(4)	94.0	0 (7)	92.8 (9)
C(1)-C(7)-C(8)	111.0	6 (7)	113-5 (1	3)
C(1)-C(7)-C(9)	114.1	7 (8)	110.4 (1	2)
C(4) - C(7) - C(8)	114.9	9 (9)	114.8 (1	3)
C(4) - C(7) - C(9)	114-	8 (8)	113.9 (1	3)
C(8) - C(7) - C(9)	106.	7 (9)	110.4 (1	0)
Br(1) - C(n) - C(7)	113.	1 (6)	111.8 (8	s)
Br(2)-C(10)-C(1)	113-4	4 (7)	113.8 (8	3)
Shortest intermolecular distance				

Br···Br	3-467 (1)	3.638 (2)

0.08 (max.) [0.13, 0.50]; max. final difference density 0.6 $[0.9] e Å^{-3}$ close to Br positions. Thermal parameters interpretable in terms of rigid-body motion; bond lengths corrected for libration. Absolute configuration from Dadson, Lam, Money & Piper (1983). Final positional parameters are in Table 1, bond lengths and angles in Table 2, and other data have been deposited.*

Related literature. The crystal structure analyses were undertaken to assist in the characterization of the products in regiospecific bromination and selective debromination synthetic routes to optically active camphor derivatives (Dadson, Lam, Money & Piper, 1983). The molecular structures (Fig. 1) are similar to those of other camphor derivatives (*e.g.* Phillips & Trotter, 1977).

* Lists of anisotropic thermal parameters, H-atom coordinates, torsion angles and structure factors have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 43043 (23 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.



Fig. 1. Stereoviews of the dibromocamphor molecules, 8,10 isomer (top) and 9,10 isomer (bottom).

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References

DADSON, W. M., LAM, M., MONEY, T. & PIPER, S. E. (1983). Can. J. Chem. 61, 343-346.

International Tables for X-ray Crystallography (1974). Vol. IV, pp. 99–102, 149. Birmingham: Kynoch Press. (Present distributor D. Reidel, Dordrecht.)

PHILLIPS, S. E. V. & TROTTER, J. (1977). Acta Cryst. B33, 200-202.

Acta Cryst. (1986). C42, 1454-1455

Structure of 9-Bromo-3-endo-methylcamphor*

BY STEVEN J. RETTIG AND JAMES TROTTER

Department of Chemistry, University of British Columbia, Vancouver, BC, Canada V6T 1Y6

(Received 6 December 1985; accepted 26 June 1986)

C(11) Br' O' C(1') C(2')

C(6') C(7') C(8') C(9') C(10') C(11')

Abstract. $C_{11}H_{17}BrO$, $M_r = 245 \cdot 16$, orthorhombic, $P2_12_12$, $a = 14 \cdot 887$ (1), $b = 21 \cdot 590$ (2), $c = 7 \cdot 0528$ (3) Å, $V = 2266 \cdot 8$ (3) Å³, Z = 8 (2 molecules per asymmetric unit), $D_x = 1 \cdot 436$ (1) g cm⁻³, Mo Ka, $\lambda \alpha_1 = 0 \cdot 70930$ Å, $\mu = 36$ cm⁻¹, F(000) = 1008, T = 295 K, R = 0.058 for 1178 reflections. The geometry and dimensions of the molecule are similar to those in related materials; the C(1)-C(7)-C(4) bridgehead angle is $92 \cdot 7$ (7)°, and the mean C-Br bond distance is $1 \cdot 98$ (1) Å.

Experimental. Colourless crystals, dimensions $0.15 \times$ 0.12×0.60 mm, faces {010}, {001}, {120}; Enraf-Nonius CAD-4F diffractometer; lattice parameters a from 25 reflections with $\theta = 15-21^{\circ}$. Intensities for $\theta \le 25^{\circ}$, *hkl*: 0 to 17, 0 to 25, 0 to 8; ω -2 θ scan, ω scan width $(0.65 + 0.35 \tan \theta)^{\circ}$ at $1.3-10^{\circ} \min^{-1}$, Br extended 25% on each side for background measure-0 ment; three standard reflections (no decay); Lp and C(1) C(2) intensity absorption corrections, transmission factors C(3) 0.38-0.57; 2304 independent reflections measured, C(4) 1178 with $I \ge 3\sigma(I), \sigma^2(I) = S + 4(B_1 + B_2) + (0.04S)^2$, C(5) C(6) $S = \text{scan}, B_1, B_2 = \text{background counts. Structure by}$ C(7) Patterson and Fourier methods; refined by full-matrix C(8) C(9) least squares on F; H atoms in calculated positions; C(10)





grams; final R = 0.058, wR = 0.076 for 1178 reflections, S = 3.12, 235 parameters; R = 0.121 for all 2304 reflections; $\Delta/\sigma = 0.06$ (mean), 0.29 (max.); max. final difference density -0.8 to +0.6 eÅ⁻³. Final positional parameters are in Table 1, bond distances Table 1. Final positional (fractional. $\times 10^4$. Br $\times 10^5$)

enantiomorph gave significantly poorer agreement

(R = 0.079 vs 0.071 before final refinement); w =

 $1/\sigma^2(F)$; scattering factors from International Tables

for X-ray Crystallography (1974); locally written, or

locally modified versions of standard computer pro-

able	1. Final pos	itional (fra	ctional, × 1	$0^{\circ}, Br \times 10^{\circ})$
nd	equivalent	isotropic	thermal	parameters
	$(U \times 10^3 \text{ Å}^2)$) with e.s.d.	's in parent.	heses

x	У	z	U_{eq}^*
36185 (15)	25692 (9)	42254 (24)	89
5701 (8)	1496 (5)	11381 (19)	157
5286 (11)	1524 (6)	8040 (19)	45
5460 (10)	1801 (8)	9919 (22)	53
5399 (9)	2478 (7)	9888 (23)	48
5155 (9)	2564 (6)	7773 (17)	40
5964 (10)	2443 (9)	6583 (22)	66
6136 (10)	1754 (8)	6840 (24)	64
4536 (9)	1993 (7)	7392 (20)	46
3665 (11)	1942 (7)	8540 (21)	62
4305 (11)	1876 (7)	5278 (21)	60
5091 (14)	826 (8)	7872 (29)	93
6199 (13)	2819 (9)	10703 (24)	93
36392 (16)	39594 (9)	106580 (24)	89
2172 (8)	5521 (6)	3693 (17)	86
2222 (9)	5189 (7)	6994 (21)	46
2605 (11)	5331 (7)	5030 (24)	54
3586 (11)	5210 (6)	5112 (17)	46
3716 (9)	5033 (6)	7193 (19)	42
3594 (14)	5597 (7)	8407 (22)	64
2618 (13)	5749 (8)	8145 (24)	65
2846 (9)	4650 (6)	7569 (17)	40
2745 (12)	4085 (7)	6350 (23)	64
2683 (10)	4507 (7)	9604 (20)	59
1258 (14)	5126 (9)	7013 (28)	89
4165 (13)	5760 (10)	4361 (29)	95

* $U_{eq} = \frac{1}{3}$ trace of diagonalized U tensor.

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